## **AMENDMENTS TO THE CLAIMS**

This listing of claims will replace all prior versions, and listings, of claims in the application:

## **Listing of Claims:**

1. (currently amended) A method for the removal of silver from a cuprous
chloride solution in a copper recovery process, comprising removing, in at least two stages,
silver from [[a]] the cuprous chloride solution with soluble mercury, using wherein fine-grained
copper powder is fed countercurrently to the cuprous chloride solution, the method comprising:
feeding mercury into the cuprous chloride solution at preselected stages in a
preselected molar ratio to the silver in the solution;
precipitating a generated silver amalgam onto [[a]] the surface of fine-grained
copper <u>powder</u> ;
removing the precipitated silver amalgam from the cuprous chloride solution for
the separation of mercury and silver;
recycling soluble mercury back to silver removal; and
treating the removed precipitated silver amalgam compound for the recovery of
silver.
2. (currently amended) [[A]] The method according to claim 1, wherein th

- 2. (currently amended) [[A]] <u>The</u> method according to claim 1, wherein the molar ratio of mercury to silver in a first amalgam precipitation stage is 0.5–2.
- 3. (currently amended) [[A]] <u>The</u> method according to claim 1, wherein the molar ratio of mercury to silver in a second amalgam precipitation stage is at least 2.
- 4. (currently amended) [[A]] The method according to claim 3, wherein the molar ratio of mercury to silver in the second amalgam precipitation stage is between 2 10.

- 5. (currently amended) [[A]] <u>The</u> method according to claim 1, wherein the particle size of the fine-grained copper powder has a particle size of is less than 200 µm.
- 6. (currently amended) [[A]] <u>The</u> method according to claim 5, wherein the amount of <u>fine-grained</u> copper powder <u>being countercurrently fed feed</u> is in the range of 100 g/L.
- 7. (currently amended) [[A]] <u>The</u> method according claim 1, further comprising feeding the <u>fine-grained</u> copper powder <u>countercurrently</u> to a mercury removal stage after <u>the at least two</u> silver removal stages, <u>from which it moves countercurrently in relation to the solution flow.</u>
- 8. (currently amended) [[A]] <u>The</u> method according to claim 1, further emprising wherein the treating step comprises leaching the precipitated silver amalgam into a dilute chloride solution using an oxidant, whereby the mercury dissolves as mercury chloride and the silver precipitates as silver chloride.
- 9. (currently amended) [[A]] <u>The</u> method according to claim 8, wherein the oxidant used is sodium hypochlorite.
- 10. (currently amended) [[A]] <u>The</u> method according to claim 8, wherein the oxidant used is hydrogen peroxide.
- 11. (currently amended) [[A]] <u>The</u> method according to claim 8, wherein the oxidant used is oxygen.
- 12. (currently amended) [[A]] <u>The</u> method according to claim 8, further comprising routing the <u>dissolved</u> mercury chloride back to <u>the silver</u> leaching <u>step</u>.
- 13. (currently amended) [[A]] <u>The</u> method according to claim 8, further comprising routing the silver chloride to <u>a</u> silver recovery <u>step</u>.

- 14. (currently amended) [[A]] <u>The</u> method according to claim 8, wherein an the alkali chloride content of the concentrated dilute chloride solution in the leaching step is at least 200 g/L.
- 15. (currently amended) [[A]] <u>The</u> method according to claim 1, wherein an amount of monovalent copper in the cuprous chloride solution to be purified is comprises 30 100 g/L of monovalent copper.
- 16. (currently amended) [[A]] The method according to claim 1, wherein silver removal is performed at a the cuprous chloride solution has pH value of 1 5 in the precipitating and removing steps.
- 17. (currently amended) [[A]] <u>The</u> method according to claim 1, further comprising removing silver from the cuprous chloride solution using fine-grained copper <u>powder</u> in a stage prior to the <u>before</u> at least two silver amalgam precipitation <u>stages</u> occurs with <u>mercury</u>.
- 18. (currently amended) [[A]] <u>The</u> method according to claim 17, wherein the particle size of the fine-grained copper powder has a particle size of is less than 200 μm.
- 19. (currently amended) [[A]] <u>The</u> method according to claim 18, wherein the amount of <u>fine-grained</u> copper powder <u>being countercurrently fed</u> feed is in the range of 100 g/L.